Molecularly Imprinted Polymer Pearls Obtained by Phase Inversion for the Selective Recognition of Hypericin

ANA-MIHAELA (FLOREA) GAVRILA*, TANTA-VERONA IORDACHE*, TEODOR SANDU*, ANAMARIA ZAHARIA, ANITA-LAURA RADU*, CATHERINE BRANGER, HUGUES BRISSET, ELENA-BIANCA STOICA, STELUTA APOSTOL, ANDREI SARBU*B

*National Research and Development Institute for Chemistry and Petrochemistry ICECHIM, Advanced Polymer Materials and Polymer Recycling, 202 Splaiul Independentei, 060021 Bucharest, Romania.

1Laboratoire MAPIEM, EA 4323, Universite de Toulon, CS 60584, 83041 Toulon Cedex 9, France

In this respect, the preparation of MIPs using wet-phase inversion is of great interest in the fields of separations and sensors due to the high economical potential, simple principle and well-established technological background [15-17]. The principle of this procedure is that a mixture of polymer and target molecules of interest is prepared in a casting solvent, and then imprinted polymer films, membranes or pearls are created. The chemical crosslinking step (usually performed as the imprinting step in MIT) is replaced by a physical crosslinking of the polymer precursors, via hydrogen bonds, obtained either by solvent evaporation (dry inversion) or by precipitation in non-solvent (wet inversion). The last stage of this procedure is the same as for all MIP preparation procedures, meaning template removal from the solid polymer structure [16].

A facile and inexpensive approach was established to prepare spherical molecularly imprinted polymers with hypericin (H-MIPs) via the wet-phase inversion method. The H-MIPs were characterized by infrared spectroscopy and thermogravimetric analysis. High-performance liquid chromatography combined with UV–Visible spectroscopy were used to assess the efficiency of the extraction procedure and, furthermore, to investigate the recognition properties of the H-MIPs. Obtaining an imprinting factor of 3.38 for hypericin strongly indicated a successful formation of molecularly imprinted cavities onto the H-MIPs. A fair selectivity towards hypericin was tested in comparison to the interfering molecule pseudohypericin.

Keywords: molecular imprinting, wet-phase inversion, pearls, hypericin

In the last years, the molecular imprinting technique (MIT) has proven to be a versatile and low-complexity chemical method for the production of highly selective artificial receptors towards given molecules (templates). MIT is bio-inspired from the well known key in the lock principle, postulating that only one key fits in the lock [1-3]. By using MIT, low cost and high stability materials (known as molecularly imprinted polymers, MIPs), containing cavities able to recognize selectively templates ranging from valuable or toxic small molecules can be easily prepared. Consequently, these attributes make MIPs ideal for replacing the extensive materials commonly used in solid-phase extraction [4,5], drug delivery [6], sensors [7,8] and purification and separation [9-11].

The template used in this work was hypericin, one of the most important bioactive metabolite isolated from St. John’s Wort plant (Hypericum perforatum L.). Belonging to naphthodianthrones (NTs) group, hypericin has gained increasing interest recently due to its potential as a highly effective anti-depressant and anti-tumor agent [12, 13]. Isolation and purification of hypericin (which coexists in very low amounts with pseudohypericin, PH- structural analogue) from the plant itself is a great challenge due to its poor solubility and degradation upon exposure to heat and light [14]. Thus, to improve the extraction process of hypericin simplified procedures that allow efficient and quantitative separation are still on high demand.

In this respect, the preparation of MIPs using wet-phase inversion is of great interest in the fields of separations and sensors due to the high economical potential, simple principle and well-established technological background [15-17]. The principle of this procedure is that a mixture of polymer and target molecules of interest is prepared in a casting solvent, and then imprinted polymer films, membranes or pearls are created. The chemical crosslinking step (usually performed as the imprinting step in MIT) is replaced by a physical crosslinking of the polymer precursors, via hydrogen bonds, obtained either by solvent evaporation (dry inversion) or by precipitation in non-solvent (wet inversion). The last stage of this procedure is the same as for all MIP preparation procedures, meaning template removal from the solid polymer structure [16].

Recently, our group applied the wet-phase inversion method to produce MIP pearls able to selectively separate hypericin from Hypericum Perforatum L. primary extracts [18,19]. By this way, hypericin molecules can be generated from concentrated NTs extracts in situ, meaning along with the phase inversion of the copolymer. Moreover, this attractive technique was adapted by our group in order to prepare 2,4,6-trinitrotoluene (TNT) MIP films used further as potential recognition elements for TNT-sensor devices [20]. To this day, few reports on hypericin-MIPs have been published in the literature.

Hence, the present work aims at exploring the possibility of using the simpler and more economic wet-phase inversion method to prepare MIP pearls with pure hypericin as template (noted H-MIPs) and methacrylic acid (MAA) and acrylonitrile (AN) as functional monomers. Applying this method, two pearls systems imprinted with either 5 wt.% or 10 wt.% hypericin and containing 20 wt.% or 25 wt.% MAA ratios were successfully prepared. The properties of the H-MIPs, such as adsorption capacity, imprinting factor and selectivity, evaluated by chromatographic techniques indicated that the obtained MIPs have great potential to be used as separation materials for NTs.

Experimental part

Materials

The acrylonitrile (AN, 98 %) and methacrylic acid (MAA, 99%) monomers used for the synthesis of copolymers were purchased from ACROS Organics. AN and MAA were purified prior to use by distillation. Dimethylsulfoxide (DMSO, p.a. grade) and ethanol (EtOH, 99.6%) used for the preparation of precursor solutions and washing, respectively, were purchased from Scharlau or Chimpopar and used without further purification. The commercial hypericin (H, 99%) used as template for the imprinting procedure was supplied by Sigma-Aldrich. The hydro-alcoholic extract (30/70, v/v) containing hypericin and its structural analog pseudohypericin (0.25 g/L NTs, from
Plantavorel Romania) was used as received. Distilled water used throughout the experiments was obtained from a laboratory purification system.

**Instrumentation**

For the structural analyses, the pearls were dried at 50 °C until constant weight (around 24 h). FTIR spectra were registered on a Fourier transform infrared spectrometer Bruker Tensor 37 Spectrometer (ATR device) in the 400-4000 cm⁻¹ range using 16 scans with a 4 cm⁻¹ resolution. Thermogravimetric analysis (TGA/DTG) was performed on a Q500 TA Instrument. The samples were heated from room temperature to 800°C at a scanning rate of 10°C/min under a constant nitrogen flow rate. All binding and selectivity experiments were performed using a HPLC (Varian Prostar, Prostar 350 autosampler) equipped with a Mediterranea SEC C18, 5µm 15 x 0.46 cm column (Teknokroma) and a diode array detector (DAD) at 593 nm. In order to detect the concentration of NTs in the 400-800 nm wavelength range and to quantify the selectivity parameters of pearls, a UV-vis Spectrometer Thermo Nicolet Evolution 500 was used.

**Preparation of MIP and NIP pearls**

The two series of hypericin-MIPs (H-MIPs), based on poly(AN-co-MAA) synthesized with either 20 or 25 wt.% MAA, where the template amount varied from 5 to 10 wt.% hypericin were prepared according to a previous reported wet-phase inversion method [19]. The pearls were noted hereafter MIP 20-5/ MIP 20-10 and MIP 25-5/ MIP 25-10. This method also describes the preparation of the AN-MAA copolymer used as precursor for the pearls matrix. For short, poly(AN-co-MAA) was synthesized according to a soap-free emulsion polymerization, using two weight ratios of AN and MAA (i.e. 80:20 and 75:25 wt.%, respectively). In light-protected flasks, MIP precursor solutions were obtained by dissolving 8 wt.% poly(AN-co-MAA) into DMSO (100 mL) at around 80°C and then the template hypericin (5 or 10 wt.% calculated to the polymer) was added at room temperature. The solutions were introduced into a syringe-like cartridge of the semi-automated dripping system that converted the copolymer droplets into polymer pearls at contact with the water coagulation bath. The phase inversion was complete after 24 h and the pearls were collected from the water bath. Extraction of hypericin from the H-MIPs, performed in order to generate the specific binding sites, was accomplished by washing the pearls several times with distilled water to remove DMSO (25 mL per portion / phase) and with ethanol (25 mL per portion / phase, 4h under continuous stirring) to remove hypericin. Finally, some pearls were dried for further characterization and rest of the pearls were kept in ethanol to prevent drying/shrinking. Non-imprinted polymer pearls (noted NIP 20-5, NIP 20-10 and NIP 25-5, NIP 25-10) were prepared in the same manner as the H-MIPs but without template and served as control samples.

**Batch binding and selectivity experiments**

600 mg of wet pearls were mixed with 3 mL of hypericin/pseudohypericin hydro-alcoholic solution or NTs phyt-extract (0.025 wt.%), properly diluted in the 10⁻¹⁰⁻¹⁰⁻⁵ mole/L range, at room temperature. After 24 h the supernatants were collected and analyzed by HPLC - equipped with an UV-vis detector (DAD). Because of their high structural resemblance, hypericin and pseudohypericin absorb light in a close wavelength range (593 and 592 nm, respectively). The imprinting effect of hypericin was determined by HPLC and the selectivity of pearls was analyzed by UV-vis spectroscopy in the presence of pseudohypericin. Conditions HPLC: mobile phase: methanol/20 mM K₂HPO₄ (pH 7.0) 90:10 adjusted to pH 5.5 with phosphoric acid; isocratic mode, 1 mL/min flow rate, 10 µL injection volume.

The amounts of adsorbed hypericin and pseudohypericin (or binding capacity Q, g compound/g pearls) were calculated by subtracting the supernatant concentrations from the initial concentrations of hypericin and pseudohypericin, respectively, in solution (eq. 1). The imprinting factor, IF, quantified the specificity with which the MIP adsorbed hypericin relative to the NIP, and was determined as the ratio between the binding capacity of the MIP and that of the NIP (eq. 2).

\[
Q = \left(\text{initial concentration} - \text{supernatant concentration}\right) * \frac{\text{volume}}{\text{polymer weight (g)}} \\
IF = \frac{Q_{\text{MIP}}}{Q_{\text{NIP}}} 
\]

The distribution coefficient \(K_d\) (mL/g) represented the amount of compound \(Cn\), bounded by 1g of pearls, relative to the amount of the same compound remaining in 1 mL of feed extract (eq. 3). Selectivity coefficient, \(k\), (eq. 4) quantifies the binding affinity of MIPs for hypericin over its competitor (pseudohypericin); where the notation \(C1-C2\) represented the hypericin-pseudohypericin. A relative selectivity coefficient, \(k'\) was also defined in eq. 5), which expresses the selectivity of MIPs in competitive adsorption relative to blank pearls-NIPs. The binding experiment was carried out in duplicate and the reproducibility was found to be ±2%.

\[
K_d = \left(\text{initial concentration} - \text{supernatant concentration}\right) * \frac{\text{volume}}{\text{polymer weight (g)}} \\
k_{\text{C1-C2}} = \frac{K_d_{\text{C1}}}{K_d_{\text{C2}}} \\
k'_{\text{C1-C2}} = \frac{k_{\text{C1}}}{k_{\text{C2}}} \\
k_{\text{C1}} = \frac{K_d_{\text{C1}}}{K_d_{\text{C2nip}}} \\
k'_{\text{C1-C2nip}} = \frac{k_{\text{C1}}}{k'_{\text{C2nip}}} 
\]

**Results and discussions**

Properties of pearls

The choice of copolymer matrices as optimum precursors for imprinting is crucial. In our previous study, we proved that both the MIP and the NIP pearls presented tails at decreasing content of MAA and of hydro-alcoholic extract extract in the feed solutions [19]. Hence, in order to achieve the formation of tailless pearls following the phase inversion, precursor copolymers with 20 and 25% MAA were used. As seen in figure 1, the addition of pure hypericin (instead of extract [19]), in the form of a hydro-alcoholic NTS extract solution, showed the same effect on the shape of the pearls, with no tails. It is worth mentioning that the imprinted pearls were rose-pink derived from the red-colored hypericin (fig. 1 a and b). Furthermore, imprinting with 10% hypericin leads to a much more pronounced dyeing of pearls (fig. 1 b). From figure 1 c and d, it can be observed that the pink colour disappears upon hypericin extraction.

Hydrodynamic measurement of the internal pore volume of imprinted H pearls, \(V_p\) (mL) revealed a high pore volume between 89-92% relative to total pearl volume, \(V_c\) (mL) (table 1). The results indicated lower values for the pore volume for NIP pearls compared to the MIP ones, probably due to the absence of the template during the formation of pearls. Moreover, pearls with a higher pore volume were obtained when the content of MAA in the copolymer was higher, i.e. 25 wt%. This result arise from the fact that the AN-MAA copolymers with higher MAA content are more hydrophobic and swell in water. Hence, imprinting with pure hypericin leads to higher pore volumes (from about 88% to...
92%) compared to values obtained in our previous studies [19, 23].

By analyzing the water resulted from the phase inversion bath it was possible to quantify the total amount of hypericin that was imprinted in the pearls (table 1). For pearls imprinted with 5 wt.% hypericin over 94% was imprinted in the pearls and for the pearls with 10 wt.% hypericin, more than 99%. This can be explained by the fact that hypericin is prone to form homo-associations of 3-5 molecules at high concentrations, which are even less soluble in water. This phenomenon was confirmed by the less successful extraction procedure for the MIP 20-10 and MIP 25-10 (41% and 56%), indicating that half of the hypericin was still trapped in the matrix after extraction.

Therefore, the process was repeated for the two types of MIP pearls until the pearls were colorless. On the other hand, more than 85% of hypericin was extracted from MIP 25-5 and around 70% from MIP 20-5. The explanation for the more efficient extraction of hypericin in this case can be correlated with the higher pore volume of these pearls, which favors the release of hypericin.

**FTIR Analysis**

To confirm the presence of the monomers and the efficiency of hypericin extraction procedure, the MIPs were analyzed using FTIR spectroscopy before and after hypericin extraction. The spectrum of pure hypericin is shown in figure 2. Hypericin exhibit typical bands of naphthodianthrones. Wide $\nu_{OH}$ bands (intramolecular hydrogen bonding) appear at 3448 cm$^{-1}$, accompanied by $\nu_{C-O}$ (Ar-OH) in the 1229-1744 cm$^{-1}$ range (where quinones have intense bands assigned to the carbonyl groups). The bands from 2926 cm$^{-1}$ and 2960 cm$^{-1}$, respectively, correspond to stretching vibrations of CH and CH$_2$ groups.

As shown in table 2, the H-MIPs before and after template extraction, show distinctive bands in the 2934-2960 cm$^{-1}$ range which can be assigned to $\nu_{CH}$ stretching vibrations from the polymer backbone. In the 2242-2252 cm$^{-1}$ region, bands corresponding to $\nu_{C=O}$ from AN were observed and the bands appearing in the 3704-3296 cm$^{-1}$ and 1714-1732 cm$^{-1}$ range were noticed.

### Table 1

<table>
<thead>
<tr>
<th>Polymer</th>
<th>$m_p$ (mg)</th>
<th>$V_w - V_p$ (mL)</th>
<th>$V_p$ (%)</th>
<th>Hypericin in bath after imprinting (%)</th>
<th>Hypericin extracted (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>MIP 20-5</td>
<td>40.6</td>
<td>0.56</td>
<td>90.17</td>
<td>6.46</td>
<td>69.65</td>
</tr>
<tr>
<td>NIP 20-5</td>
<td>50.7</td>
<td>0.51</td>
<td>82.10</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>MIP 20-10</td>
<td>43.1</td>
<td>0.56</td>
<td>89.77</td>
<td>0.90</td>
<td>41.22</td>
</tr>
<tr>
<td>NIP 20-10</td>
<td>55.7</td>
<td>0.50</td>
<td>81.29</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>MIP 25-5</td>
<td>40.4</td>
<td>0.56</td>
<td>91.74</td>
<td>2.99</td>
<td>84.84</td>
</tr>
<tr>
<td>NIP 25-5</td>
<td>58.5</td>
<td>0.54</td>
<td>88.39</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>MIP 25-10</td>
<td>44.4</td>
<td>0.56</td>
<td>90.69</td>
<td>0.66</td>
<td>56.45</td>
</tr>
<tr>
<td>NIP 25-10</td>
<td>62.6</td>
<td>0.54</td>
<td>87.72</td>
<td>-</td>
<td>-</td>
</tr>
</tbody>
</table>

1. $m_p$, the weight of dried pearls (600 mg of wet sample).
2. $V_w$, the water volume determined thermo-gravimetrically (values of water volume were equivalent to the pore volume $V_p$).
3. $V_p$, the mean pore volume calculated relative to the hydrodynamic volume of wet pearls.
4. The initial amount of hypericin for MIP 20-5 and MIP 25-5 and MIP 20-10 and MIP 25-10 is 0.16 mg and 0.32 mg, respectively; the amount of water in the inversion bath was 1L.

### Table 2

<table>
<thead>
<tr>
<th>Polymer</th>
<th>$v_{\text{COOH}}$ (C=O)</th>
<th>$v_{\text{COOH}}$ (-COOH)</th>
<th>$v_{\text{ArOH}}$ (Ar-OH)</th>
<th>$v_{\text{ArCOOH}}$ (Ar-COOH)</th>
</tr>
</thead>
<tbody>
<tr>
<td>MIP 20-5 be</td>
<td>3818-3856</td>
<td>2244</td>
<td>1714</td>
<td>1450</td>
</tr>
<tr>
<td>MIP 20-5 se</td>
<td>3758-3800</td>
<td>2244</td>
<td>1720</td>
<td>1450</td>
</tr>
<tr>
<td>MIP 20-10 be</td>
<td>3896-3978</td>
<td>2244</td>
<td>1729</td>
<td>1431</td>
</tr>
<tr>
<td>MIP 20-10 se</td>
<td>3922-3952</td>
<td>2244</td>
<td>1745</td>
<td>1455</td>
</tr>
<tr>
<td>MIP 25-5 be</td>
<td>3462-3540</td>
<td>2244</td>
<td>1720</td>
<td>1435</td>
</tr>
<tr>
<td>MIP 25-5 se</td>
<td>3542-3590</td>
<td>2244</td>
<td>1727</td>
<td>1435</td>
</tr>
<tr>
<td>MIP 25-10 be</td>
<td>3370-3410</td>
<td>2244</td>
<td>1732</td>
<td>1435</td>
</tr>
<tr>
<td>MIP 25-10 se</td>
<td>3422-3452</td>
<td>2244</td>
<td>1735</td>
<td>1435</td>
</tr>
</tbody>
</table>

* Before extraction, ** after extraction.
cm⁻¹ range were characteristic for the stretching vibrations of -OH and -C=O groups, respectively, from the carboxyl functionality of MAA. It is relevant to point out that the bands corresponding to the aromatic -OH groups overlap of hypericin with the -OH groups of MAA. Hence, the imprinting effect was mostly highlighted by the shifting of ν\_OH (from MAA) due to interactions with hypericin. Polymer-template interactions were also distinguished in the H-MIPs before hypericin extraction, as summarized in table 2. As for H-MIPs pearls with a higher amount of hypericin (10%), there is a shift of bands to higher wavelengths due to the formation of homoassociates of 3-5 molecules of hypericin which prevent the establishment of stronger interactions within the polymeric matrix. The extraction of hypericin from the MIPs (see MIPs after extraction) was confirmed by the absence of major characteristic band of hypericin for the quinoid form of naphthodianthrones, at around 1360 cm⁻¹ (C-O stretching vibration). FTIR results also validate the efficiency of the template extraction procedure.

**Thermal properties**

The obtained results indicate that the imprinted pearls obtained by phase inversion are relative thermally stable (fig. 3. and fig. 4.). Thermal degradation does not occur completely as a result of the cyclization phenomenon, which is characteristic for polymers with high contents of AN (-CN pendant groups. Thus, for both series of MIPs after extraction of hypericin, residues up to 38% are formed (fig. 3. and fig. 4. a). For the pearls before hypericin extraction, a lower residue amount was noticed due the steric hindrance of hypericin that prevented cyclization of -CN fragments. The 10 °C movement of the maximum cyclization temperature confirms this hypothesis.

The decomposition of statistical copolymer chain takes place in several stages (fig. 3. b). Due to the occurrence of template-intramolecular hypericin bonds, the imprinted pearls are more stable and exhibit a more homogeneous degradation of the polymer, as indicated by the less noticeable shoulder (shifted to higher degradation temperatures). Curves recorded for MIP 20 pearls after extraction exhibited lower degradation rates than those of non-extracted pearls. The weight loss around 172 °C may be assigned to hypericin degradation (demethylation), as hypericin is thermo-sensitive. Then, a new thermal degradation step occurs, with a peak at about 250 °C (clear cyclization peak). The last decomposition step for the MIP 20-5 has a maximum at 389 °C (before hypericin extraction) and 403 °C (after hypericin extraction). This last stage shows in both cases a shoulder around 320 °C, which is more intense for the extracted pearls.

The thermal degradation of MIP 25-5 (fig. 4. b) is similar to that of MIP 20-5. The only difference was related to lower temperature maxima recorded for each decomposition step. This behavior may be due to the increase in the MAA content from 20 to 25%, which implicitly leads to a quantitative increase of -COOH groups and less residue. It was also observed that the MIPs imprinted with 10% hypericin exhibited lower weight losses compared to those with 5% hypericin.

**Imprinting studies of polymers**

The imprinting effect was quantified for hypericin. Re-binding properties of hypericin demonstrated a strong imprinting effect on all MIP pearls, these pearls adsorbing hypericin nearly three times more than the corresponding NIPs (table 3). High values of binding capacities were recorded for the pearls synthesized with 5% hypericin, i.e. MIP 20-5 and MIP 25-5. The maximum value for the binding capacity recorded for MIP 25-5, 2.65 mg/g was relative high considering the high content of MAA in the polymer matrix and the low content of hypericin used in the synthesis. This effect was also observed when imprinting with a NTs extract [19]. Yet, herein the use of pure hypericin led to higher values of the adsorption capacities. This behaviour also correlates well with the recorded pore volume for the H-MIP pearls (which was with 4% higher than the NTs-MIP pearls obtained in the previous study [19]). Furthermore, the difference of binding affinities for hypericin, between the H-MIP and the control NIP, clearly indicates the effect of imprinting with the formation of specific binding sites for hypericin.

The H-MIP pearls exhibited higher affinities for hypericin, than the non-imprinted ones leading to maximum
imprinting factor, IF, of 3.38 for MIP 25-5 (table 3). As expected, the imprinting factor showed high values for all imprinted pearls due to the imprinting effect. However, the adsorption for hypericin was found to be higher for pearls obtained with a lower amount of template, i.e. MIP x-5. Specific cavities were formed when intermolecular interactions were minimal, i.e. MIP 25-5. The theory of molecular imprinting explains this behavior referring to the concentration effect that prevents the formation of homogeneous cavities in terms of binding energy. Thus, higher amounts of template can generate a larger number of cavities, but their binding energies are different because the template molecules interact with each other during imprinting, which leads to different conformations; therefore, cavities have different levels of energy.

In the selectivity experiment, the target molecule hypericin and the competitive one pseudohypericin possess similar structure and co-exist in Hypericum extract as bioactive components. Selectivity of MIPs towards pseudohypericin was studied by contacting the imprinted and non-imprinted pearls in contact with the NTs extract, which contained around 40% hypericin and 60% pseudohypericin [19]. The parameters of the selectivity experiments are shown in Table 3. The Hypericin extract contains more pseudohypericin than hypericin and, as a result, the distribution coefficient, K, values for pseudohypericin are higher compared to those for hypericin. Based on the values of K, there was no significant difference of rebinding hypericin and pseudohypericin with NIPs (k = 0.73 and 0.91). However, MIPs present close values of adsorption selectivities (k) for hypericin (maximum value of 1.03 for MIP 25-10). The low range of k values was in agreement with the very similar quinone structures of hypericin and pseudohypericin. Moreover, it can also be observed that the selectivity was influenced by the MAA content in the polymer matrix. Overall, the relative selectivity coefficients k' confirm that the selectivity of hypericin-MIPs was higher than that of corresponding NIPs which suggested the imprinting effect with hypericin.

Conclusions

In conclusion, pearl shaped MIPs with reproducible properties were synthesized via wet phase inversion using MAA as functional monomer and AN as structural monomer and pure hypericin as the template. The structure and thermal stability of the resulted MIPs were characterized using FTIR and TGA. Hydrodynamic measurements of the hypericin-MIP pearls revealed a dense porous structure with high pore volumes of about 92 vol.%. The extraction procedure of hypericin from the MIP pearls was more efficient for the MIP x-5 pearls due to their high pore volume that favored a quantitative extraction. The polymer with the highest content of MAA i.e. MIP 25-5 showed high adsorption capacities (Q = 2.65 mg g⁻¹) but fair selectivity towards hypericin against pseudohypericin, compared to our previous paper [19, 23]. However, the value of the imprinting factor (IF = 3.38) recorded for MIP 25-5 was in agreement with the relative selectivity coefficients, k’. Overall, the reported strategy for preparing hypericin-MIP pearls is efficient and the use of pure hypericin in the imprinting step leads to good imprinting parameters of polymers. Therefore, the adsorbent materials can be used to obtain more concentrated hypericin solutions.

Acknowledgments

The authors thank for the help and support of the Romanian Ministry of Research and Innovation (MCI) and the Executive Unit for Financing Higher Education, Research, Development and Innovation (UEFISCDI) through Project NUCLEU PN.19.23.02.01. and also the EU in the frame of the collaborative international consortium (39/2017-ProWsper) financed under the ERA-NET Cofund WaterWorks2015 Call.

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Manuscript received: 8.04.2019